

Cambridge Assessment International Education

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/62

Paper 6 Alternative to Practical

October/November 2017

MARK SCHEME
Maximum Mark: 60

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2017 series for most Cambridge IGCSE[®], Cambridge International A and AS Level components and some Cambridge O Level components.

 ${\rm \rlap{R}\hskip-1pt B}$ IGCSE is a registered trademark.

Cambridge Assessment
International Education

[Turn over

© UCLES 2017

0653/62

Cambridge IGCSE – Mark Scheme **PUBLISHED**

| Question | Answer | Marks |
|-----------|---|-------|
| 1(a) | quality of drawing ; | 3 |
| | larger than original; | |
| | root and stem correctly labelled ; | |
| 1(b)(i) | 14 ± 1 ; | 1 |
| 1(b)(ii) | correct measurement (in mm) ; | 1 |
| 1(b)(iii) | magnification correctly calculated and rounded correctly; | 1 |
| 1(c)(i) | (placed in a suitable container with) water ; | 2 |
| | kept in a warm place ; | |
| 1(c)(ii) | protein ; | 2 |
| | starch; | |

| Question | Answer | Marks |
|-----------|--|-------|
| 2(a)(i) | Different coloured ppts.; | 2 |
| | same coloured ppts. as NaOH or ammonia ; | |
| 2(a)(ii) | (add H to) iron(II) sulfate (solution); | 1 |
| 2(a)(iii) | add aqueous sodium hydroxide (and heat) ; | 2 |
| | (in gas) red litmus turns blue / pungent smell ; | |
| 2(b)(i) | add (dil) HCl or (dil) nitric acid; | 2 |
| | bubbles / effervescence ; | |

0653/62

Cambridge IGCSE – Mark Scheme **PUBLISHED**

| Question | Answer | Marks |
|-----------|--|-------|
| 2(b)(ii) | test-tube / flask connected to test-tube / beaker with delivery tube ; | 2 |
| | limewater in correct place labelled ; | |
| 2(b)(iii) | (limewater goes) milky/white ppt.; | 1 |

| Question | Answer | Marks |
|----------|--|-------|
| 3(a)(i) | 89 <u>.0 ;</u> | 1 |
| 3(a)(ii) | 92.5 ; | 1 |
| 3(b)(i) | s / second / sec and °C; | 1 |
| 3(b)(ii) | 30, 60, 90, 120, 150, 180 ; | 1 |
| 3(c) | to allow maximum temperature of hot water to be recorded / let thermometer get to temperature of water / beakers owtte ; | 1 |
| 3(d) | rate of temperature drop greater at start / wtte ; | 1 |
| 3(e) | use of table results (e.g. 89 to 80 and 92.5 to 76); | 2 |
| | reference to temperature changes in the same time / same time specified; | |
| 3(f) | Any two from: room temperature / initial water temperature / same volumes of water / keep thermometer the same depth ;; | 2 |

© UCLES 2017 Page 3 of 5

Cambridge IGCSE – Mark Scheme **PUBLISHED**

| Question | Answer | Marks |
|-----------|---|-------|
| 4(a)(i) | 1.5 and 4.9; | 1 |
| 4(a)(ii) | 3.4 3.0 all 3 correct is 2 marks ;; 1 or 2 correct is 1 mark | 2 |
| 4(a)(iii) | Low 0.16 and High 0.32 | 1 |
| 4(a)(iv) | Increase in temperature increases rate / ORA ; | 1 |
| 4(b) | Any two from: Identify anomalies / more reliable / average is more accurate ;; | 2 |
| 4(c)(i) | transpiration; | 1 |
| 4(c)(ii) | Water stored or produced or used by plant ; | 1 |
| 4(d) | (reduced) because surface area reduced / fewer stomata / less evaporation of water / less transpiration AW; | 1 |

| Question | Answer | Marks |
|-----------|--|-------|
| 5(a)(i) | sulfuric; | 1 |
| 5(a)(ii) | added solid does not dissolve / no more bubbles ; | 1 |
| 5(a)(iii) | (blue) litmus ; does not go red ; | 2 |
| 5(b)(i) | funnel with filter paper drawn ; | 2 |
| | (excess) copper carbonate and filtrate / copper sulfate labelled ; | |

0653/62

Cambridge IGCSE – Mark Scheme **PUBLISHED**

| (| October/Nove | ember |
|---|--------------|-------|
| | | 2017 |

| Question | Answer | Marks |
|----------|--|-------|
| 5(b)(ii) | insoluble (in water); | 1 |
| 5(c) | to avoid dehydration / to avoid decomposition; | 1 |
| 5(d)(i) | sodium hydroxide is soluble ; | 1 |
| 5(d)(ii) | use an indicator for neutralisation ; | 1 |

| Question | Answer | Marks |
|-----------|--|-------|
| 6(a) | 66.2 ;; | 2 |
| 6(b) | 6.2 and 9.0 ; | 1 |
| 6(c)(i) | suitable choice of scales (≽ half the grid used) ; | 3 |
| | plots correct to half a small square, at least 4 correct; | |
| | good best fit line judgement ; | |
| 6(c)(ii) | indication on graph of how data obtained AND at least half of line used; | 2 |
| | correct calculation for triangle method using data from graph ; | |
| 6(c)(iii) | correct answer using candidate's gradient ; | 1 |
| 6(d) | rule will not balance / balance point is off the scale of the rule ; | 1 |

© UCLES 2017 Page 5 of 5